# Behavior of the <sup>222</sup>Rn daughters on the copper surface during cleaning

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#### Outlook

- Technique
- Testing cleaning procedures
  - etching
  - electropolishing
- Comparing etching with electropolishingConclusions

#### Why to investigate <sup>222</sup>Rn daughters?

- □ Equilibrium in the chain broken at the <sup>210</sup>Pb level
- <sup>210</sup>Pb may stay as a main residual contamination of the copper after cleaning (will appear after some years)
- <sup>210</sup>Po radio-chemistry not well understood
- Long-lived <sup>222</sup>Rn/<sup>210</sup>Pb daughters implanted/deposited on the copper surface may contribute to the background in GERDA



#### Why to investigate <sup>222</sup>Rn daughters?



#### Technique

- Screening of <sup>210</sup>Po with an alpha spectrometer
  50 mm Si-detector, bcg ~ 5 α/d (1-10 MeV)
  sensitivity ~ 20 mBq/m<sup>2</sup> (50 mBq/kg, <sup>210</sup>Po Ag)
- Screening of <sup>210</sup>Bi with a beta spectrometer
  2×50 mm Si(Li)-detectors, bcg ~ 0.14/0.26 cpm sensitivity ~ 6 Bq/kg (<sup>210</sup>Bi 0.012/0.023 cpm/(Bq/kg))
- □ Screening of <sup>210</sup>Pb (46.6 keV line) with a gamma spectrometer

25 % - n-type HPGe detector with an active and a passive shield, sensitivity  $\sim 20~Bq/kg$ 

Only small samples can be handled – artificial contamination needed: copper discs loaded with <sup>222</sup>Rn daughters

#### Technique

- LENS electrolytic copper used to fabricate sample discs (50 mm diameter, 1 mm thickness)
- □ Discs cleaned applying "Majorana procedure" (5 min in 1%  $H_2SO_4 + 3\% H_2O_2$ ; 5 min in 1% citric acid; rinsing with distilled water)
- Discs placed for 4 months in a strong <sup>222</sup>Rn source (1.4 MBq)

# Technique

Discs before and after cleaning



#### Discs loaded with <sup>222</sup>Rn daughters



### Outlook

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- "Majorana procedure" tested:
  copper discs 5 min in 1% H<sub>2</sub>SO<sub>4</sub> + 3% H<sub>2</sub>O<sub>2</sub>
  5 min in 1% citric acid rinsing with water
- Solution volume: each time 250 cm<sup>3</sup>
- Temperature: 20 °C
- <sup>210</sup>Po measured on both disc sides
- Blank signal:  $(0.0042 \pm 0.0005)$  cpm
- Disc loaded with <sup>210</sup>Po: side a:  $(2.97 \pm 0.03)$  cpm side b:  $(2.64 \pm 0.03)$  cpm

Loaded disc, side a:  $(2.97 \pm 0.03)$  cpm side b:  $(2.64 \pm 0.03)$  cpm

Run No.	Disc side	<sup>210</sup> Po activity [cpm]	<sup>210</sup> Po reduction factor R	Amount of removed Cu	Remarks
1	a	$2.36\pm0.03$	1.3	$(1.77 \pm 0.02) \text{ mg/cm}^2$	Acid mixed during etching
1	b	$2.16\pm0.02$	1.2	1.98 µm	
	a	$1.83 \pm 0.04$	1.3	$(2.29 \pm 0.02) \text{ mg/cm}^2$	Acid mixed during etching
2	b	$1.79\pm0.03$	1.2	2.56 µm	
	a	$1.84\pm0.03$	0.99	$(4.40 \pm 0.02) \text{ mg/cm}^2$ $4.91 \ \mu\text{m}$	Acid mixed during etching
3	b	$1.62\pm0.03$	1.1		
4	a	$1.65\pm0.03$	1.1	$(3.21 \pm 0.02) \text{ mg/cm}^2$	Acid mixed during etching
4	b	$1.43\pm0.02$	1.1	3.58 µm	
_	a	$1.62\pm0.03$	1.0	$(3.38 \pm 0.02) \text{ mg/cm}^2$	Acid mixed during etching
5	b	$1.35 \pm 0.02$	1.1	3.77 µm	
	a	$1.47\pm0.02$	1.1	$(3.47 \pm 0.02) \text{ mg/cm}^2$	Acid mixed during etching
6	b	$1.25\pm0.03$	1.1	3.87 µm	
7	a	$1.50 \pm 0.02$	0.98	$(2.37 \pm 0.02) \text{ mg/cm}^2$	Acid mixed during etching
7	b	$1.26\pm0.03$	0.99	2.64 μm	<sup>209</sup> Po added (1.42 Bq)



#### Etching (with <sup>209</sup>Po), disc No. 2



#### Results for <sup>210</sup>Pb, <sup>210</sup>Bi and <sup>210</sup>Po:

Isotope	Original activity [cpm]	Activity after cleaning [cpm]	Reduction factor R	Amount of removed Cu	Remarks
<sup>210</sup> Pb	$1.49\pm0.04$	< 0.022	> 68		Only side a was investigated
<sup>210</sup> Bi	$31.17 \pm 0.71$	$0.77\pm0.02$	40.5	3.91 mg/cm <sup>2</sup> 4.4 μm	Only side a was investigated
<sup>210</sup> Po	$2.55\pm0.01$	$2.06\pm0.01$	1.2		Only side a was investigated

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- □ Both disc sides investigated separately for <sup>210</sup>Po
- $\square \quad \text{Electrolyte: 85 \% H}_3\text{PO}_4 + 5 \% \text{ 1-butanol} (C_4\text{H}_{10}\text{O})$
- □ Only one cathode (Cu disc)
- Several runs performed, each time using a new cathode and fresh solution



U = 1.8 VI = 150 - 10 mA

#### Results:

Loaded disc side a:  $(9.52 \pm 0.06)$  cpm side b:  $(1.78 \pm 0.04)$  cpm

Run No.	Disc side	<sup>210</sup> Po activity [cpm]	<sup>210</sup> Po reduction factor R	Amount of removed Cu*	Remarks
1	a	$0.50 \pm 0.03$	19		Polished for 35 min
1	b	$1.38\pm0.03$	1.3		Total charge: 70 mAh
2	a $0.062 \pm 0.003$ 8	8	$17 \text{ mg/cm}^2$	Polished for 35 min	
2	b	$0.74 \pm 0.01$	1.9		Total charge: 70 mAh
3	a	$0.024 \pm 0.002$	2.6		Polished for 35 min
	b	$0.017 \pm 0.002$	44		Total charge: 70 mAh

- disc side facing the cathode

\*) measured after all runs

After	а	$0.024\pm0.002$	397	≤ 12.6 µm	Polished for 70 min
all	b	$0.017\pm0.002$	105	$\leq 6.3 \ \mu m$	Polished for 35 min

- □ Both sides investigated separately for <sup>210</sup>Po
- $\square \quad \text{Electrolyte: 85 \% H}_3\text{PO}_4 + 5 \% \text{ 1-butanol} (C_4\text{H}_{10}\text{O})$
- □ Only one cathode (Cu disc)
- □ One run performed, disc was turned around several times
- □ Total polishing time: 3 h

Results:

Disc side	Original <sup>210</sup> Po activity [cpm]	<sup>210</sup> Po activity after pol. [cpm]	<sup>210</sup> Po reduction factor R	Amount of removed Cu	Remarks
а	$2.18\pm0.02$	$0.011\pm0.001$	198	$20 \text{ mg/cm}^2$	Facing the cathode 3 times, each time for 30 min
b	$2.45 \pm 0.03$	$0.014 \pm 0.001$	175	22.3 µm	Facing the cathode 3 times, each time for 30 min

- □ Both sides investigated separately for <sup>210</sup>Po, <sup>210</sup>Bi and <sup>210</sup>Pb
- $\square \quad \text{Electrolyte: 85 \% H}_{3}\text{PO}_{4} + 5 \% \text{ 1-butanol} (C_{4}\text{H}_{10}\text{O})$
- □ Only one cathode (Cu disc)
- □ One run performed, disc was turned around every 5 min.
- □ Total polishing time: 1 h

#### **Results:**

Disc side	<sup>210</sup> Po [cpm] before/after	<sup>210</sup> Bi [cpm] before/after	<sup>210</sup> Pb [cpm] before/after	Amount of removed Cu	Remarks
a				$4.5 \text{ mg/cm}^2$	Facing the cathode 6 times, each time for 5 min
b	$5.31 \pm 0.12$ $0.18 \pm 0.01$ R = 30	$36.55 \pm 0.47$ $0.15 \pm 0.01$ R = 244	$2.08 \pm 0.02$ $0.002 \pm 0.002$ R = 1040	5.2 μm	Facing the cathode 6 times, each time for 5 min

# <sup>210</sup>Pb energy spectrum, disc No. 8



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#### Comparing etching with electropolishing

□ Amount of removed material:

- after 7 "Majorana" runs (30 min): 20.9 mg/cm<sup>2</sup>
- after one polishing run (35 min): 5.7 mg/cm<sup>2</sup>
- □ Amount of removed <sup>210</sup>Po activity:
  - after 7 "Majorana" runs (35 min, 20.9 mg/cm<sup>2</sup>):  $R_{av} = 2$

 $R_{av} = 30$ 

- after polishing (1 h, 4.5 mg/cm<sup>2</sup>):
- after long-polishing run (3 h, 20 mg/cm<sup>2</sup>):  $R_{av} = 187$
- □ Amount of removed <sup>210</sup>Pb and <sup>210</sup>Bi activity:
  - one "Majorana" run (5 min, 3 mg/cm<sup>2</sup>):  $R_{Bi} = 40$ ,  $R_{Pb} > 68$
  - electropolishing (1 h, 4.5 mg/cm<sup>2</sup>):  $R_{Bi} = 240$ ,  $R_{Pb} = 1000$

#### Conclusions

- Etching and electropolishing remove up to 20 mg/cm<sup>2</sup>
  Cu (depending on the exposure time)
- <sup>210</sup>Po deposited on- or close to the copper discs surface (relatively narrow **Q**-peaks)
- □ Etching does not remove <sup>210</sup>Po from the copper re-deposition (see test with <sup>209</sup>Po)
- □ Long electropolishing reduces <sup>210</sup>Po activity by a factor of ~200 much more effective than etching
- $\Box \quad \text{Etching removes most of } ^{210}\text{Pb and } ^{210}\text{Bi} (> 98 \%)$
- □ Electropolishing removes <sup>210</sup>Pb and <sup>210</sup>Bi more effective than etching (99.5 % <sup>210</sup>Bi and > 99.9 % <sup>210</sup>Pb removed)